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sphere of homogeneous solution. 6.2

## EXPLAINING SOLUTIONS

PRACTICE (Page 273) Understanding Concepts 1. Most molecules are polar.

The four categories of polar molecules are:  $AB$  (e.g.,  $HCl(g)$ ,  $CO(g)$ )  $N_xA_y$  (e.g.,  $NH_3(g)$ ,  $NF_3(g)$ )  $O_xA_y$  (e.g.,  $H_2O(l)$ ,  $OCI_2(g)$ )  $C_xA_yB_z$  (e.g.,  $CHCl_3(l)$ ,  $C_2H_5OH(l)$ ) 2.

## 6.1 DEFINING A SOLUTION

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