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liquids throughout the day as yo u

Chapter 13: States of Matter

Chapter 13 States of Matter 137 SECTION 13.1 THE NATURE OF GASES (pages 385 – 389) This section introduces the kinetic theory and describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance. Kinetic Theory and a Model for Gases (pages 385 – 386) 1.

Name Date Class STATES OF MATTER 13

There are three states of matter that we will learn about in this chapter. (If you want to learn about more

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states of matter, I can refer you to somebody.) Those three states are solid, liquid, and gas. These three states are quite different. The main difference is in their particles.

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Chapter 13 States Of Matter Chapter 13 States of Matter137 SECTION 13.1 THE NATURE OF GASES (pages 385 – 389) This section introduces the kinetic theory and describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance.

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Chapter 13: States of Matter. STUDY. PLAY. Kinetic Molecular Theory. Explains the properties of gases in terms of the energy, size, and motion of their particles. Elastic Collision. Describes a collision in which kinetic energy may be transferred between colliding particles but the total kinetic energy of the two particles remains the same.

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Authors: Wilbraham, ISBN-10: 0132525763, ISBN-13: 978-0-13252-576-3, Publisher: Prentice Hall

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all matter consists of tiny particles that are constantly in motion What are the three assumptions of the kinetic theory as it applies to gases? -The particles in a gas are considered to be small, hard spheres with an insignificant volume. -The motion of the particles in a gas are rapid, constant, and random.

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The Sustainable Development Goals are a call for action

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by all countries – poor, rich and middle-income – to promote prosperity while protecting the planet. They recognize that ending poverty ...

United Nations Sustainable Development – 17 Goals to ...

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of Matter 2 Kinetic Theory as Applied to Gases

Fundamental assumptions about gases. The particles in a gas are considered to be small, hard spheres with an insignificant volume. Between particles in a gas there is empty space. No attractive or repulsive forces exist between the particles. 3

Chapter 13 States Of Matter Worksheet

The attacks the USPS continues to face are not just attacks on the postal service but attacks on Black lives. To defund the USPS would be to deny future generations this opportunity and dishonor the legacy of Black postal workers. Now, we 're taking this matter into our own hands by writing and sending

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#BlackLoveLetters through USPS...

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